

Total Iron Vacu-vials® Kit

K-6023: 0 - 2.50 ppm (Prog. # 102)

Instrument Set-up

For CHEMetrics photometers, follow the **Setup and Measurement Procedures** in the operator's manual. For spectrophotometers, set the wavelength to 560 nm. A sealed ZERO ampoule is supplied in this kit for zeroing when the sample is colorless and not turbid. For improved accuracy with colored or turbid samples, Sample Zeroing Accessory Pack, Cat. # A-0503 is recommended. Using the sample cup, snap the tip of the A-0503 ampoule in the sample (see figure 3 below). Invert the ampoule to mix. Dry the ampoule and use it in place of the supplied ZERO ampoule to zero the instrument.

Test Procedure

1. Fill the sample cup to the 25 mL mark with the sample to be tested (fig 1).
2. Add 5 drops of A-6000 Activator Solution (fig 2). Stir to mix the contents of the cup.
3. Wait **1 minute**.
4. Place the Vacu-vial ampoule, tip first, into the sample cup. Snap the tip. The ampoule will fill leaving a bubble for mixing (fig 3).
5. To mix the ampoule, invert it several times, allowing the bubble to travel from end to end.
6. Dry the ampoule. Obtain a test result **5 minutes** after snapping tip.
7. Insert the Vacu-vial ampoule into the photometer, flat end first, and obtain a reading in ppm (mg/Liter) iron (Fe).

NOTE: If using a spectrophotometer that is not pre-calibrated for CHEMetrics products, then use the **equation below** or the **Concentration Calculator** found under the Support tab at www.chemetrics.com.

$$\text{ppm} = 2.57 (\text{abs}) - 0.02$$

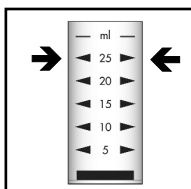


Figure 1

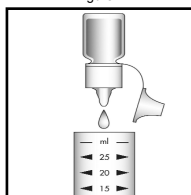


Figure 2

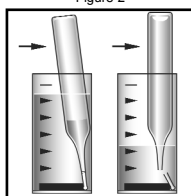


Figure 3

Test Method

The Total Iron Vacu-vials®¹ test kit employs the PDTS chemistry.^{2,3} The sample is treated with a mixture of thioglycolic acid and ammonia. This mixture dissolves most forms of particulate iron. The resulting ferrous iron then reacts with PDTS (3-(2-pyridyl)-5,6-bis(4-phenylsulfonic acid)-1,2,4-triazine disodium salt) to form a pink-purple colored complex in direct proportion to the total iron concentration. Various metals will produce high test results. Certain forms of very insoluble iron (magnetite, ferrite, etc.) require the following digestion procedure prior to running the test procedure:

- a. Fill a heat-resistant, glass container to 25 mL with the sample to be tested.
- b. Add 5 drops of A-6000 Activator solution. Stir briefly.
- c. Gently boil the sample to reduce volume to 10-15 mL.
- d. Cool the sample and dilute to 25 mL with iron-free water.
- e. Using this pretreated sample, perform the **Test Procedure**, beginning with Step 4.

1. Vacu-vials is a registered trademark of CHEMetrics, Inc. U.S. Patent No. 3,634,038
2. G. Frederick Smith Chemical Co., The Iron Reagent, 3rd ed., p. 47 (1980).
3. J.A. Tetlow and A.L. Wilson, "The Absorptometric Determination of Iron in Boiler Feed-water," Analyst. Vol. 89, p 442 (1964).

Safety Information

Read SDS (available at www.chemetrics.com) before performing this test procedure. Wear safety glasses and protective gloves.

Visit www.chemetrics.com to view product demonstration videos.
Always follow the test procedure above to perform a test.



Simplicity in Water Analysis

www.chemetrics.com

4295 Catlett Road, Midland, VA 22728 U.S.A.

Phone: (800) 356-3072; Fax: (540) 788-4856

E-Mail: orders@chemetrics.com

Sept. 19, Rev. 16